

Fuel Energy Content

The energy content of traditional carriers (methane, CR2032 button cell, Lead-acid battery, Li-ion battery) and alternative sources (hydrogen, urea, urine). Comparison per mass [Wh/kg] and volume [Wh/l].

Common Energy Sources

Methane CH₄

- Source: <https://en.wikipedia.org/wiki/Methane>
- Source: https://people.wou.edu/~courtina/GS361/Energy_From_Fossil_Fuels.htm
- 0.657 g/l (liquid at 116 K would be 442.8 g/l)
- 16.043 g/mol
- 810 kJ/mol
- $810 \text{ kJ/mol} / 16.043 \text{ g/mol} = 50.489 \text{ kJ/g} = 14025 \text{ Wh/kg} = 14 \text{ Wh/g}$
- Link 2 says: 51.6 kJ/g
- $0.657 \text{ g/l} * 14.025 \text{ Wh/g} = 9 \text{ Wh/l}$

Button cell CR2032

- Source: <https://www.digchip.com/datasheets/parts/datasheet/915/CR2032.php>
- $3 \text{ V} * 240 \text{ mAh} = 0.72 \text{ Wh} = 0.00072 \text{ kWh}$
- $\pi/4 * 0.20 \text{ dm} * 0.20 \text{ dm} * 0.032 \text{ dm} = .0010053096 \text{ l} = 1 \text{ cm}^3$
- $3 \text{ g} = 0.003 \text{ kg}$
- $0.72 \text{ Wh} / 0.003 \text{ kg} = 240 \text{ Wh/kg}$
- $0.72 \text{ Wh} / 0.0010053096 \text{ l} = 716 \text{ Wh/l}$

Lead-acid battery

- Source: <https://batteryguy.com/files/technical-docs/tech-spec-sheets/bg-121000nb-technical-specification-sheet.pdf>
- $20 \text{ h} * 5 \text{ A} * 10.8 \text{ V} = 1080 \text{ Wh} = 1.08 \text{ kWh}$
- 28.2 kg
- $3.30 \text{ dm} * 1.71 \text{ dm} * 2.14 \text{ dm} = 12.06 \text{ l}$
- $1080 \text{ Wh} / 28.2 \text{ kg} = 38 \text{ Wh/kg}$
- $1080 \text{ Wh} / 12.06 \text{ l} = 90 \text{ Wh/l}$

Li-ion battery

- Source: <https://emergoplus.com/product/powerxtreme-x10/>
- $12.8 \text{ V} * 10 \text{ Ah} = 128 \text{ Wh} = 0.128 \text{ kWh}$
- 2.1 kg
- $182 \text{ mm} * 120 \text{ mm} * 90 \text{ mm} = 1.82 \text{ dm} * 1.20 \text{ dm} * 0.90 \text{ dm} = 1.9656 \text{ l}$
- $128 \text{ Wh} / 2.1 \text{ kg} = 61 \text{ Wh/kg}$
- $128 \text{ Wh} / 1.9656 \text{ l} = 65 \text{ Wh/l}$

Alternative Energy Sources

Hydrogen H2 (gas)

- Source: <https://en.wikipedia.org/wiki/Hydrogen>
- Source: <https://www.convertunits.com/molar mass/Hydrogen>
- Source: https://people.wou.edu/~courtina/GS361/Energy_From_Fossil_Fuels.htm
- 0.08988 g/l
- 1.00794 g/mol
- 482 kJ/mol
- Link 3 says: 120 kJ/g
- $482 \text{ kJ/mol} / 1.00794 \text{ g/mol} = 478 \text{ kJ/g} = 133 \text{ Wh/g} = 133 \text{ kWh/kg}$
- $0.08988 \text{ g/l} / 1000 \text{ g/kg} = 0.00008988 \text{ kg/l}$
- $133 \text{ kWh/kg} * 0.00008988 \text{ kg/l} = 0.012 \text{ kWh/l} = 12 \text{ Wh/l}$

Urea

- Source: <https://en.wikipedia.org/wiki/Urea>
- 1.32 g/cm³
- 60.056 g/mol
- 837 kJ/mol = 232.5 Wh/mol
- $232.5 \text{ Wh/mol} / 60.056 \text{ g/mol} = 3871 \text{ Wh/kg} = 4 \text{ Wh/g}$
- $3.871 \text{ Wh/g} * 1.32 \text{ g/cm}^3 = 5.11 \text{ Wh/cm}^3 = 5110 \text{ Wh/l} = 5 \text{ kWh/l}$

Urine

- Source: <https://en.wikipedia.org/wiki/Urine>
- Source: <https://en.wikipedia.org/wiki/Urea>
- urine 1.5 kg/person/day
- urea 30 g/person/day
- concentration urea / urine = $30 \text{ g} / 1500 \text{ g} = 0.02 = 2\%$
- 2% of 3871 Wh/kg = 77 Wh/kg
- 2% of 5110 kWh/l = 102 Wh/l

Summary

Carrier	[Wh/kg]	[Wh/l]
Methane	14025	9
CR2032	240	716
Lead acid battery	38	90
Li-ion battery	51	65
Hydrogen gas	133000	12
Urea	3871	5110
Urine	77	102